

CHEM 307 B1/B2 COURSE SCHEDULE Summer 2020

Date	Reading & Learning Objectives to target per lecture*	End o/Chapter Problems
5/26	<p>Reading: Book sections: 1.1-2, 2.1-2.2</p> <ul style="list-style-type: none"> Syllabus; introduction to organic chemistry, Atomic structure and the periodic table; valence electrons Lewis structures; bonds as shared electron pairs, non-bonding electrons; common Bonding patterns for C, N, O, X; formal charge; condensed structures; line-angle structures; use of R to denote substituents Pauling electronegativity and polar bonds; <p>Recitations first day to practice Canvas Conferences-</p>	<p>Chapter 1: 18-20 Chapter 2: 18, 23</p>
5/27	<p>Reading Book sections 2.2-2.4, 10.2a-c</p> <ul style="list-style-type: none"> Introduction to resonance structures Atomic orbitals (s, p); molecular orbital theory; bonding and anti-bonding orbitals, sigma (σ) and pi (π) orbitals; bond strengths VSEPR, bond angles, molecular shape and molecular dipoles; wedge-dash notations Using molecular models Orbital hybridization and molecular shape; double and triple bonds Delocalized electrons, resonance structures and orbital hybridization <p>Practice Lecture Quiz</p>	<p>Chapter 2: 2, 16, 17, 20, 21, 22, 24, 26, 27</p>
5/28	<p>Reading Book sections: 2.5-2.7, 1.1b, 1.3-4</p> <ul style="list-style-type: none"> Writing resonance structures practice Introduction to IUPAC nomenclature and functional groups: with focus on alkanes, cycloalkanes, alkyl halides and alcohols; constitutional isomers 	<p>Chapter 1: 21a, 26, 28, 29 abce, 31, 38a Chapter: 2: 28, 29, 30, 31, 32</p>
5/30	WEEK 1 QUIZ DUE (Chapters 1-2)	
6/1	<ul style="list-style-type: none"> Reading Book sections: 2.8, 3.1-3.2 Intermolecular forces: van der Waals interactions, electrostatic forces, hydrogen bonds Physical properties and intermolecular forces and its effect on life Conformations of acyclic hydrocarbons; Newman projections; gauche, anti, eclipsed conformations; hyperconjugation as a stabilizing effect; intro to potential energy diagrams; emphasis on line-angle condensed structures; use of wedges and dashes <p>WEEK 1 HW DUE 6/1/20</p>	<p>Chapter 3: 13, 15, 16, 17, and 22</p> <p>Practice drawing cyclohexane chair conformers (p. 84)</p>

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6/2	<p><i>Reading Assignment Book sections: 3.2-3.4 and 4.1-4.2</i></p> <ul style="list-style-type: none"> • Conformations of cyclic hydrocarbons • Conformations of cyclohexane and substituted cyclohexanes; drawing chair conformers; • Equilibria between chair conformers; free energy differences and equilibrium constants • Configurational isomers: cis and trans-substituted cycloalkanes 	<p>Chapter 3: 18a,c,d; 19, 20, 22, 24, 25, and 27</p> <p>Chapter 4: 14-21, 22, 25, 26, 28-30</p>
6/3	<p><i>Reading Assignment Book sections: 4.2-4.4</i></p> <ul style="list-style-type: none"> • Stereoisomers; geometric isomers; cis- and trans-alkenes, alkene nomenclature and E/Z nomenclature. • Chiral centers, enantiomers and R/S nomenclature; wedge-dash and Fischer representations • 	<p>Chapter 4: 31, 34, 35, 36</p>
6/4	<p><i>Reading Assignment Book sections: 4.4, 5.2-5.4</i></p> <ul style="list-style-type: none"> • Optical rotation; absolute configurations: L-amino acids and D-sugars • Molecules with two or more chiral centers; diastereomers and meso compounds • Exam 1 review 	
6/7	WEEK 2 QUIZ DUE (open date 6/6/20)	
Monday, 6/8 EXAM I		
6/9	<p><i>Reading Assignment Book sections: 5.2-5.4</i></p> <ul style="list-style-type: none"> • Acid-base reactions; use of the arrow notation to indicate electron movement • pKa values: utility to indicate acid and base strengths and Keq Factors affecting acid/base strength: ion size, electronegativity, resonance, inductive effects, orbital hybridization (the arguments of organic chemistry) • Lewis acids and Lewis bases • Reaction Coordinate Diagrams 	<p>Chapter 5: 21-24, 26-29, 31, 33</p> <p>Chapter 6: 14 a-d, 15, 16-20, 21, 22 a b, 23-25</p>
6/10	<p><i>Reading Assignment Book sections: 6.1-6.2</i></p> <ul style="list-style-type: none"> • Nucleophilic substitution reactions: a mechanistic overview; energy diagrams • The SN2 reaction: mechanism (a HOMO-LUMO view), inversion of configuration; leaving groups, nucleophile strength, steric factors, solvent effects • The SN1 reaction: mechanism, carbocation intermediates 	

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6/11	<p>Reading Assignment Book sections: 6.2-6.4, and 7.1a-c; 2d</p> <ul style="list-style-type: none"> • SN2 (a HOMO-LUMO view), inversion of configuration; leaving groups, nucleophile strength, steric factors, solvent effects • The SN1 reaction: mechanism, carbocation intermediates and carbocation stability (hyperconjugation), racemic products; solvent effects; carbocation rearrangements • SN2 vs. SN1: competing reactions under kinetic control • Substitution reactions of alcohols and ethers in acid: generation of a good leaving group by protonation 	<p>Chapter 6: 14 a-d, 15, 16-20, 21, 22 a b, 23-25</p>
6/15	<p>Reading Assignment Book sections: 7.2a-c, e</p> <ul style="list-style-type: none"> • Substitution reactions of alcohols using alkylsulfonyl chlorides (RSO₂Cl); conversion to alkyl halides using phosphorous tribromide or thionyl chloride • Synthesis of ethers and epoxides • Substitution reactions of epoxides in acid and base 	<p>Chapter 7: 17, 23, 25, 27d, 29a,b,d-g</p>
6/13-14	<p>WEEK 3 QUIZ</p>	
6/16	<p>Reading Assignment Book sections: 8.1-8. 3</p> <ul style="list-style-type: none"> • Elimination reactions: a mechanistic overview, energy diagrams, alkene stability • E2 reactions of alkyl halides and sulfonate esters; mechanism, stereoselectivity and regioselectivity; competition with SN2 reactions • E1 reactions of alcohols in acid 	<p>Chapter 8: 13, 15-17, 20-22, 23 a-d, f-j, 25, 26 a b d e f i j; 28 a c d e</p>
6/17	<p>Reading Assignment Book sections: 9.1-9.2</p> <ul style="list-style-type: none"> • E1 reactions of alkyl halides; competition with SN1 reactions Substitution versus elimination: alkyl halides, alcohols • Electrophilic addition to alkenes: addition of HX, H₂O • 	<p>Chapter 9: 23 – 28, 29 a b d g, 30</p>
6/18	<p>WEEK 3 HW DUE</p>	
6/18	<p>Reading Assignment Book sections: 9.4- 9.5</p> <ul style="list-style-type: none"> • Electrophilic addition of Cl₂, Br₂ to alkenes: stereoselectivity, regioselectivity; • Halohydrin formation • Oxymercuration/demercuration: regioselectivity • Hydroboration/oxidation: stereoselectivity, regioselectivity • Alkene polymerization 	<p>Chapter 9: 25e h, 29 ce, 32 b c, 33, 34</p>
6/20	<p>Exam review</p>	
<p>Monday, 6/22 EXAM II</p>		

Date	Reading & Learning Objectives to target per lecture*	Recommended Problems
6/23	<p>Reading Assignment Book sections: 10.1-10.3</p> <ul style="list-style-type: none"> • Electrophilic addition to alkynes • Carbenes: synthesis of cyclopropane rings • Multi-step synthesis using substitution, elimination and addition reactions: strategies, retrosynthetic analysis Dienes and polyenes: structure, nomenclature and stabilities • Molecular orbital descriptions of conjugated dienes and polyenes; UV-VIS absorption by conjugated polyenes, colored organic compounds • Electrophilic addition to conjugated dienes: kinetic <i>vs</i> thermodynamic control 	<p>Chapter 10: 11-14, 16, 18-21, 23 a e</p>
6/23	Week 4 Quiz Due	
6/24	<p>Reading Assignment Book sections: 10.4, 11.1-11.3</p> <ul style="list-style-type: none"> • The Diels-Alder reaction: mechanism, MO description and HOMO-LUMO theory, • stereoselectivity and regioselectivity • Oxidation states in organic chemistry • Catalytic hydrogenation reactions: reduction of alkenes and alkynes • Oxidation reactions of alkenes • Synthesis 	<p>Chapter 11: 10, 11, 13-16, 19</p>
6/25	<p>Reading Assignment Book sections: 11.4; 12.1 -</p> <ul style="list-style-type: none"> • Oxidation reactions of alcohols • Nomenclature of carbonyl compounds • Multi-step synthesis and retrosynthetic analysis • Free radical halogenations: mechanism; bond energies and regioselectivities; • the Hammond postulate; selectivity of bromination <i>vs</i> chlorination 	<p>Chapter 1: 22a,b,d,f; 30b,c,d,g,h, 32, 34 Chapter 11: 12d-d, 17, 23, 27b</p>
6/28	WEK 5 QUIZ due	
6/29	<p>Reading Assignment Book sections: 12.2a-c, 3a-b</p> <ul style="list-style-type: none"> • Reductions of benzyl alcohols, ethers and esters via radical intermediates: stability of the benzyl radical • Alkyne reduction by sodium in liquid ammonia: mechanism, stereoselectivity • Free radical addition reactions: anti-Markovnikov HBr addition, alkene polymerization 	<p>Chapter 12: 17, 18, 19g,h; 21a, 22a, 23a, 24a, 27 d,i, j, 28</p>
6/30	<p>Reading Assignment Book sections: 12.4ac Oxidation <i>via</i> radical intermediates: fatty acid oxidation, vitamins C and E, anti oxidant food preservatives, anti-oxidants in colored fruits</p>	<p>Chapter 12: 19, 25, 27, 28, 30a, 33, 34</p>
7/1	Final Exam Review	
<p>Thursday, 7/2 FINAL EXAM (comprehensive w/emphasis on materials not covered in exams 1 and 2) *depending on class progress and time some parts may be skipped</p>		